

## FLAVONOIDS OF *PLAGIOCHILA ASPLENIoidES*

RUEDIGER MUES and H. DIETMAR ZINSMEISTER

Fachbereich 16-Botanik, Universität des Saarlandes, 6600 Saarbrücken, GFR

(Received 29 April 1976)

**Key Word Index**—*Plagiochila asplenoides*; Jungermaniales; Hepaticae; liverwort; flavones; di-C-glycosides of apigenin, luteolin, tricetin and tricetin-5'-methyl ether.

**Abstract**—Besides an apigenin- and a luteolin-di-C-glycoside, 5 previously unknown di-C-glycosides of tricetin were identified in the gametophytic and sporophytic tissues of *Plagiochila asplenoides*. Two of them were new 6,8-di-C-hexopyranosyltricetins, and two were new 6-C-hexopyranosyl-8-C-pentopyranosyltricetins. 6-C-Hexopyranosyl-8-C-pentopyranosyltricetin-5'-methyl ether was also found.

### INTRODUCTION

In the last 10 years the occurrence of flavonoids in liverworts (Hepaticae) has become well established [1-5]. Whereas the order Marchantiales has been studied in detail, mainly by Markham and coworkers [6-13], the order Jungermaniales has until now been largely ignored. We started some years ago an extensive study of European Jungermaniales; 60 species of 35 genera belonging to 23 families—about 80% of those occurring in Europe—have been so far included. In 18 species of 11 genera and 11 families flavonoids could be detected. They were provisionally identified from their chromatographic data and behaviour on acid hydrolysis to be flavone O- and C-glycosides [14].

One of the investigated species which has flavonoids is *Plagiochila asplenoides*. Glycoflavones in this species were previously mentioned by Harborne [15] and in a previous communication we reported the occurrence of luteolin-di-C-glycosides [16]. The present paper revises this communication and reports additionally the isolation of a variety of tricetin-di-C-glycosides and an apigenin-di-C-glycoside (vicenin).

### RESULTS

From the extract of 1 kg air-dried mixed gametophytic and sporophytic plant material 8 different flavone glycosides (compounds 1-8) could be isolated by combined column-PC. The chromatographic and UV-spectral data are given in Table 1; the MS-data of the permethylated (PM) and perdeuteriomethylated (PDM) derivatives according to Brimacombe *et al.* [17] are found in Table 2. The yields were compound 1 (42 mg), 2 (5 mg), 3 (5 mg), 4 (16 mg), 5 (8.5 mg), 6 (tr), 7 (7 mg) and 8 (2 mg).

**Compound 8.** This was the only substance clearly separated from the other compounds which appeared on a 2-D chromatogram as a complex streaking spot with different concentration centers after spraying with Naturstoffreagenz A [16] (NA). The deep purple absorbance in UV (254, 350 nm), the fluorescence after spraying with NA and Benedicts reagent and the UV-spectral data indicated an apigenin derivative. After acid treatment no

aglycone or free sugar was produced, but after PC an additional spot close to the original one appeared as a consequence of Wessely-Moser-rearrangement [18] of a C-glycoside. The MS showed a  $M^+$  at  $m/e$  748 and at  $m/e$  781 corresponding with the MW of PM- and PDM-di-C-hexosylapigenins respectively. The MS-data of the PM-derivative are in agreement with those of Bouillant *et al.* [19, 20] for 6,8-di-C-hexosylapigenin (vicenin). However, it did not correspond with vicenin 2 (6,8-di-C- $\beta$ -D-glucopyranosylapigenin) after co-chromatography with an authentic sample and from this finding it is concluded that compound 8 is a 6,8-di-C-hexosylapigenin with different hexoses at C-6 and C-8. Further structural confirmation of the sugar-moiety was impossible because of the small quantity of the compound isolated.

**Compound 5.** The chromatographic behaviour was different from that of compound 8 and its UV-spectrum (Table 1) suggested that it was a flavone glycoside of the luteolin type. Acid treatment indicated C-glycosidation and the  $M^+$  at  $m/e$  778 and  $m/e$  814 corresponded with the MW of PM- and PDM-di-C-hexosylluteolins respectively. This is confirmed by the absence of pentose fragment peaks [19, 20]. The periodate  $-NaBH_4$  degradation product was glycerol [21]. Compound 5 was thus concluded to be 6-8-di-C-hexopyranosylluteolin. After acid isomerization no additional spot was observed, showing that the presence of the same sugar moieties at the C-6 and C-8 position. Co-chromatography with natural 6-8-di-C- $\beta$ -D-glucopyranosylluteolin (lucenin 2) was identical in four different solvents (Table 1).

Compounds 1-4 and 7 showed very similar chromatographic behaviour to compound 5 (lucenin). Also the UV-spectra were nearly identical, with one difference that the Band II region [22] of the NaOMe spectrum of these compounds showed only shoulders or very weak peaks. Nevertheless the similarity of the UV-spectra with those of lucenins led in a previous communication [16] to the conclusion that they are also luteolin-6,8-di-C-glycosides. However, the correct interpretation of their MS which was first possible by the extensive studies of Chopin's group [19, 20] led us to the conclusion that these compounds are derivatives of 5'-hydroxyluteolin (tricetin).



C-6 position and the pentose to the C-8 position. The periodate  $\text{-NaBH}_4$  degradation products for both compounds were glycerol and ethyleneglycol derived from an aldohexose in a pyranose form and a pentose in a pyranose form respectively. The occurrence of different sugars at the C-6 and C-8-position is supported by the results of acid isomerization where additional spots were observed. From all these data it is concluded, that compounds 2 and 3 are both 6-C-hexopyranosyl-8-C-pentopyranosyl-tricetins but contain different stereoisomers of the sugar moieties.

**Compound 7.** The experimental data (of Tables 1 and 2) led to the conclusion that this compound must also be a 6,8-di-C-glycosyl-tricetin of a similar type to compounds 2 and 3. The  $M^+$  of the PM-derivative was found to be the same as those of compounds 2 and 3, but the  $M^+$  of the PDM-derivative was at  $m/e$  797, compared with those of the PDM derivatives of compounds 2 and 3 at  $m/e$  800; this is due to one free OH group less in the underivatized molecule than in compounds 2 and 3. Further experiments also indicated that the 5' (or 3')-position is methoxylated. The results of acid isomerization and periodate  $\text{-NaBH}_4$  degradation were similar to those of compound 3. Compound 7 is therefore a 6-C-hexopyranosyl-8-C-pentopyranosyltricetin-5' (or 3')-methylether.

Compound 6 was isolated in such a small quantity that it was not possible to run MS of its derivatives. From the chromatographic, UV-spectral and acid isomerization data it could possibly be another tricetin-di-C-glycoside with different sugars at the C-6 and C-8 positions.

To our knowledge this is the first report of tricetin-di-C-glycosides in plants. Until now only a few reports of the occurrence of O-glycosides or sulphates of tricetin have appeared [24-26]. Our work has also shown that flavone-C-glycosides are more common in liverworts than was previously supposed. This is supported by the fact that most flavonoid containing species of Jungermanniales, investigated in this laboratory, seem to have C-glycosides confirming the earlier results based on acid treatment. Besides the occurrence of the previously reported di-C-glycosides of the apigenin- and luteolin-type in liverworts [3, 4, 8, 12] we have found a third flavone aglycone type with di-C-glycosidation. These tricetin-C-glycosides are very unstable in the presence of oxygen, like luteolin-di-C-glycosides [27]. They decompose or polymerize to products, which are soluble only in water giving a bright yellow colour. After development with standard solvents for flavonoids they cannot be detected on PC in UV or after spraying with the usual reagents for flavonoids.

#### EXPERIMENTAL

**Plant material.** *Plagiochila asplenioides* (L.) Dum. was collected in June, 1974 near Klagenfurt, Austria. Voucher specimens are deposited in the Herbarium of the Fachrichtung Botanik, Universität des Saarlandes, Saarbrücken.

**Isolation procedure.** Air-dried plant material was extracted as described previously [16]. Individual compounds were isolated by column chromatography on cellulose followed by PC (often "overrun" to achieve separation) on Whatman 3 MM paper. Purity of compounds was monitored on TLC (microcrystalline cellulose and polyamide). Solvents: column 5% up to 40% HOAc; PC (see Table 1); TLC-microcrystalline cellulose (see Table 1), polyamide—20 and 50% MeOH. The com-

pounds did not crystallise and were lyophilised. Chromatographic and UV-spectral identification of the compounds was carried out as previously described [16].

**Acid isomerization.** Compounds (1-2 mg) were treated with 5 ml 2 N HCl under reflux at 100° for 1½-2 hr.

**Periodate  $\text{-NaBH}_4$  degradation** was performed according to ref. [21]; detection of the degradation products was done by the method of ref. [28].

**PMR.** Flavone glycoside (10 mg) dissolved in DMSO  $d_6$  and deuterated TFA, TMS internal standard; 60 MHz.

**MS.** Preparation of PM- and PDM-derivatives was performed according to ref. [17].

**Acknowledgements.**—We are indebted to Prof. Chopin and Miss Bouillant, Université Claude Bernard, Lyon, for running MS of compound 1 and helpful discussion and for an authentic sample of natural lucenin 2. For running the MS of the other compounds and the PMR of compound 1 and valuable advices for interpretation we thank Mr. O. Seligmann and Prof. Wagner, Inst. für Pharmazeutische Arzneimittellehre, Universität München. Technical assistance of Miss S. Klein is gratefully acknowledged.

#### REFERENCES

1. Reznik, H. and Wiermann, R. (1966) *Naturwissenschaften* **53**, 230.
2. Nilsson, E. (1969) *Acta Chem. Scand.* **23**, 2910.
3. Nilsson, E. (1973) *Phytochemistry* **12**, 722.
4. Markham, K. R., Porter, L. J. and Brehm, B. G. (1969) *Phytochemistry* **8**, 2193.
5. Tjukavkina, N. A., Benešová, V. and Herout, V. (1970) *Coll. Czechosl. Chem. Commun.* **35**, 1306.
6. Markham, K. R. (1972) *Phytochemistry* **11**, 2047.
7. Markham, K. R., Mabry, T. J. and Averett, J. E. (1972) *Phytochemistry* **11**, 2875.
8. Markham, K. R. and Porter, L. J. (1973) *Phytochemistry* **12**, 2007.
9. Markham, K. R. and Porter, L. J. (1974) *Phytochemistry* **13**, 1553.
10. Markham, K. R. and Porter, L. J. (1974) *Phytochemistry* **13**, 1937.
11. Markham, K. R. and Porter, L. J. (1975) *Phytochemistry* **14**, 199.
12. Markham, K. R., Porter, L. J., Mues, R., Zinsmeister, H. D. and Brehm, B. G. (1976) *Phytochemistry* **15**, 147.
13. Markham, K. R., Porter, L. J. and Miller, N. G. (1976) *Phytochemistry* **15**, 151.
14. Mues, R. (1975) Diss. Saarbrücken.
15. Harborne, J. B. (1966) in *Comparative Phytochemistry* (Swain, T., ed.), p. 271. Academic Press, New York.
16. Mues, R. and Zinsmeister, H. D. (1975) *Phytochemistry* **14**, 577.
17. Brimacombe, J. S., Jones, B. D., Stacey, M. and Willard, J. J. (1966) *Carbohydr. Res.* **2**, 167.
18. Wessely, F. and Moser, G. H. (1930) *Monatsh. Chem.* **56**, 97.
19. Bouillant, M.-L., Favre-Bonvin, J. and Chopin, J. (1974) *C.R. Acad. Sci. Paris Sér. D*, **279**, 295.
20. Bouillant, M.-L., Favre-Bonvin, J. and Chopin, J. (1975) *Phytochemistry* **14**, 2267.
21. Viscontini, M., Hoch, D. and Karrer, P. (1955) *Helv. Chim. Acta* **38**, 642.
22. Mabry, T. J., Markham, K. R. and Thomas, M. B. (1970) *The Systematic Identification of Flavonoids*, Springer, Berlin.
23. Chopin, J. and Bouillant, M.-L. (1975) in *The Flavonoids* (Harborne, J. B., Mabry, T. J., and Mabry, H., eds.), p. 632. Chapman & Hall, London.
24. Beckmann, S. and Geiger, H. (1968) *Phytochemistry* **7**, 1667.

25. Lamer, E. and Bodalski, T. (1968) *Diss. Pharm. Pharmacol.* **20**, 623.
26. Williams, C. A., Harborne, J. B. and Crosby, T. S. (1976) *Phytochemistry* **15**, 349.
27. Seikel, M. K., Chow, J. H. S. and Feldman, L. (1966) *Phytochemistry* **5**, 439.
29. Galle, K. (1974) *Diss. München*.